

SPLIT SAMPLING EVENT DATA REVIEW AND DISCUSSION

Data collected at the Homestake Mining Company Grants site by USGS and HMC/Arcadis in 2016

February 15, 2018



Meeting Agenda

- Introductions
- Site Orientation
- Split Sampling
 - Big picture conclusions
 - HMC-USGS data comparisons
 - Data by sampling method
- Passive Samplers
- Geophysics
- New DD wells
- Drilling, soil sampling, and geophysics at DD and DD-2
- Action Items



Health and Safety Moment



Safety vest for personal vehicular use

- Being struck by a vehicle is the sixth leading cause of occupation-related death for California Highway Patrol officers
- Keep a vest in your vehicle, preferably within reach without having to exit vehicle
 - In glove compartment
 - Under front seat
 - In pocket behind seat



2 Options Available

3M High-Visibility Yellow Reflective Personal Safety Vest

Model# 94616-80030



\$1097

✓ Free shipping with \$45 order

34 in stock to pick up today Check nearby stores

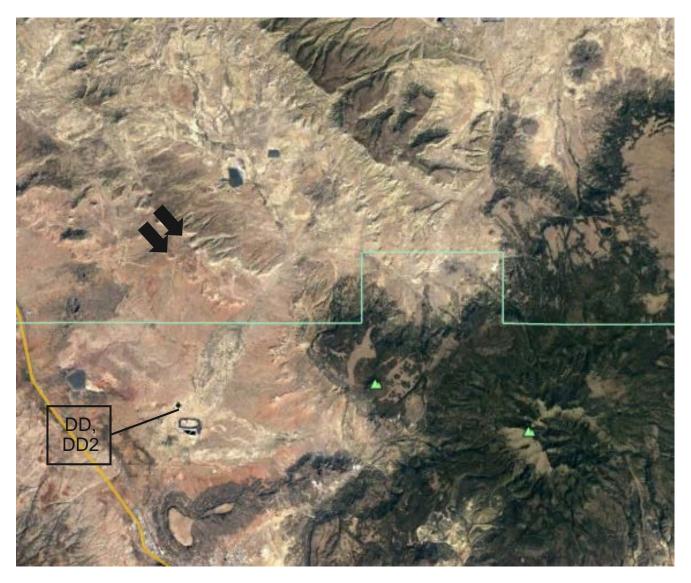
Add to Cart



The site

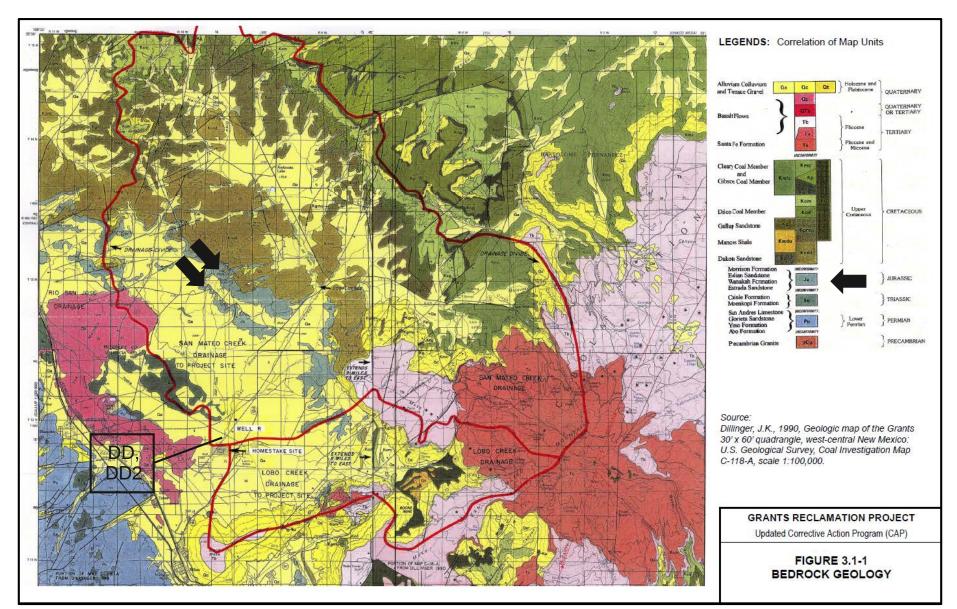
San Mateo Creek Basin





San Mateo Creek Basin Geology ARCADIS Design & Consultancy for natural and built assets





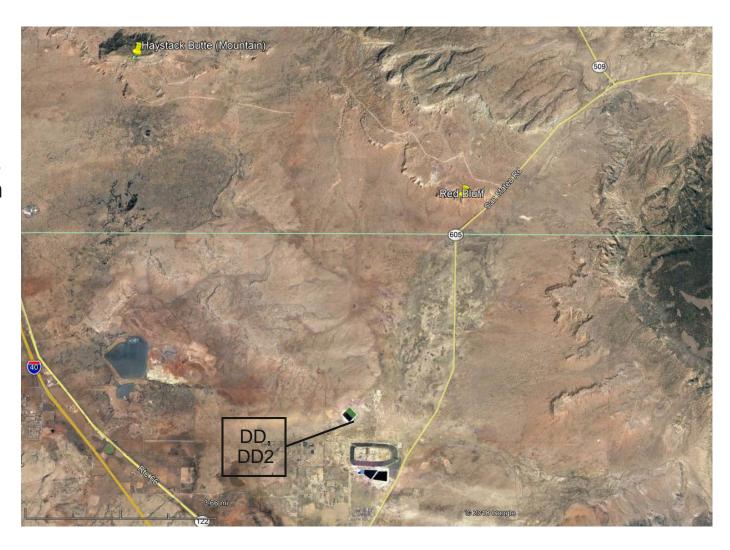


Alluvium comes from eroded highlands

This rock contains ore-grade uranium

Results in disseminated uranium particles in alluvium

Erosion/fluvial deposition is a heterogeneous process (visible)





Split sampling event: Summer 2016



What was collected

- Field parameters
- 3 types of water samples: volumetric, micropurge, passive sampler
- Metals
- Major anions and cations
- Nitrogen compounds
- Alkalinity
- Total organic carbon
- Radionuclides
- Isotopes
- Dissolved gases (CFCs)
- Geophysical data
- Field Hach analyses: dissolved oxygen and ferrous iron



What we have received

- ✓ Field parameters
- ☐ 3 types of water samples: volumetric, micropurge, passive sampler
- ✓ Metals
- ✓ Major anions and cations
- ✓ Nitrogen compounds
- ✓ Alkalinity
- ✓ Total organic carbon
- □ Radionuclides
- ✓ Isotopes
- ✓ Dissolved gases (CFCs)
- ✓ Geophysical data
- ✓ Field Hach analyses: dissolved oxygen and ferrous iron



Split Sampling data – comparing the splits



HMC-USGS Data Comparison

USGS Sulfate Stable Sulfur Isotopes (%)

15

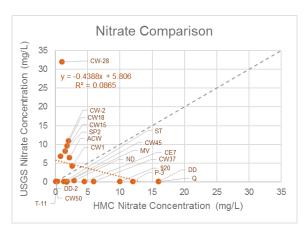
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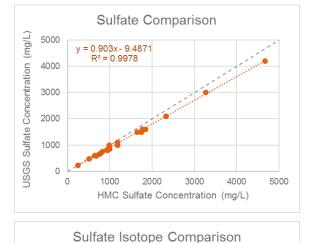
-5

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-35



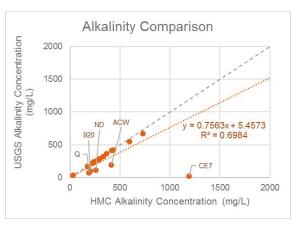


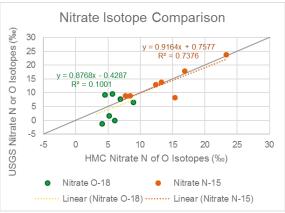
y = 0.9428x - 1.0585

 $R^2 = 0.9927$

-5

15





Reasonable data comparability most samples

lower alkalinity concentrations

Five USGS samples with substantially

Reasonable data comparability

Conclusions similar for both data sets

– nitrate primarily from sewage /
manure with nitrate reduction
occurring

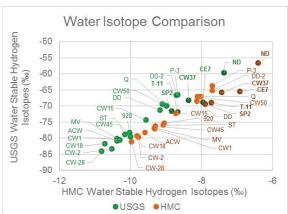
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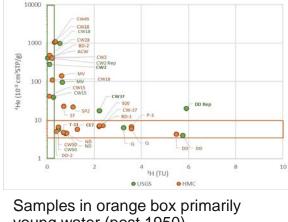
-15

HMC Sulfate Stable Sulfur Isotopes (%)

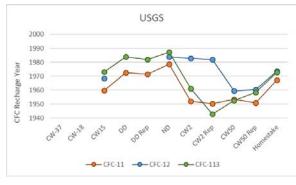


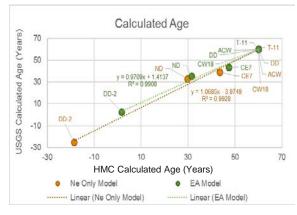
HMC-USGS Data Comparison





3H and 4He

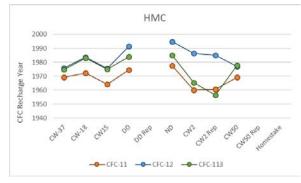






Samples in green box primarily old water (prior to 1950) - water "age" increases with increasing ⁴He concentration

Samples outside of boxes are likely mixtures of young and old water, except USGS DD-Rep and CW37 sample results which were qualified as poor fit



HMC recharge year younger than USGS recharge year for CW-15, CW-2. and CW-50

Poor correlation between recharge vear for the different CFCs for both USGS and HMC data sets

HMC data heavier than USGS

Although HMC data heavier than USGS, interpretations consistent across the two data sets - SP2, T-11, CW37, and CE7 are more evaporated than other samples



HMC-USGS Data Comparison

Poor correlation (R<0.25):

Ammonia nitrogen

Antimony D,T

Cadmium D

Cobalt D,T

Chromium D,T

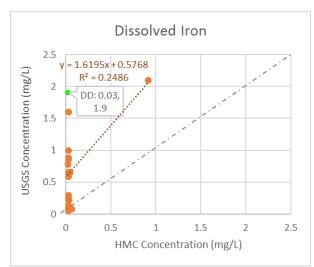
Iron D,T

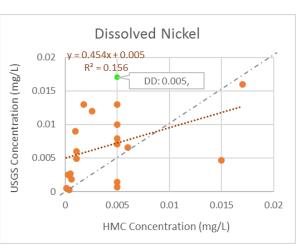
Lead D,T

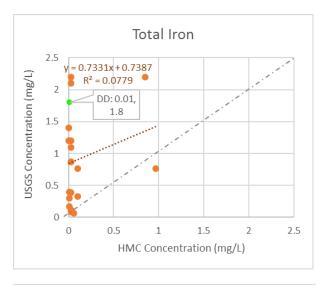
Nickel D,T

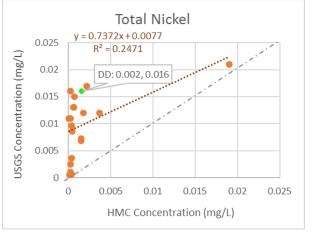
Zinc D

Specific conductivity, Ra-228 (T), and potassium (T) show moderate-high correlation after removal of one outlier











Split sampling data and major water chemistries

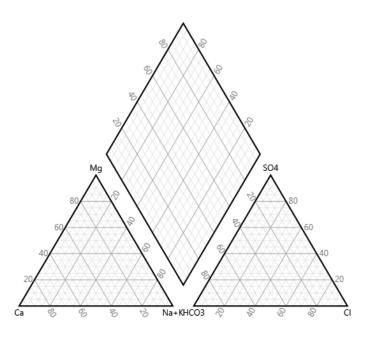
Types of plots and diagrams



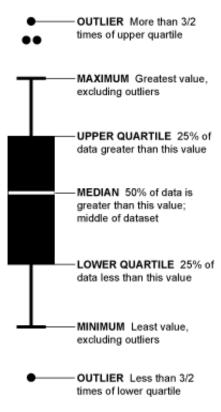
Stiff diagram

Cations Anions meq/I 20 10 10 20 Na+K CI Ca HCO3+CO3

Piper diagram



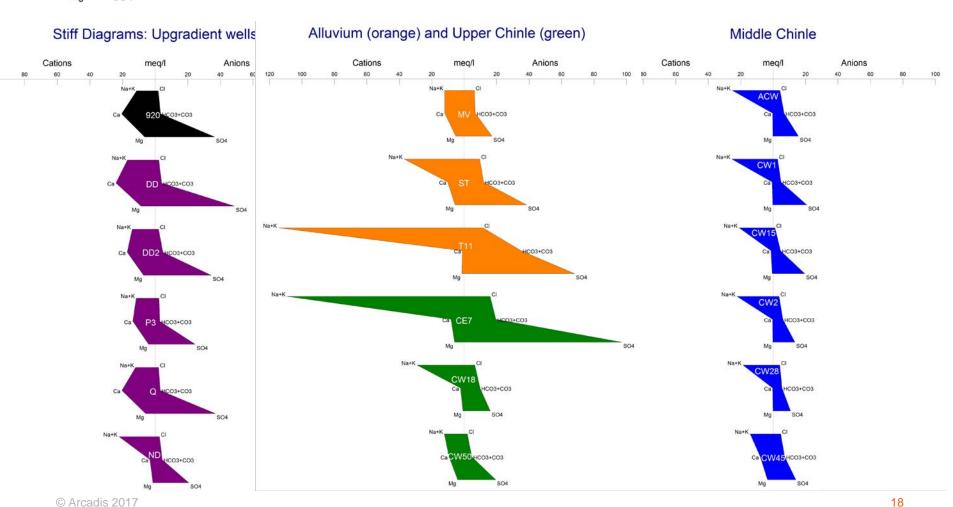
Box plot



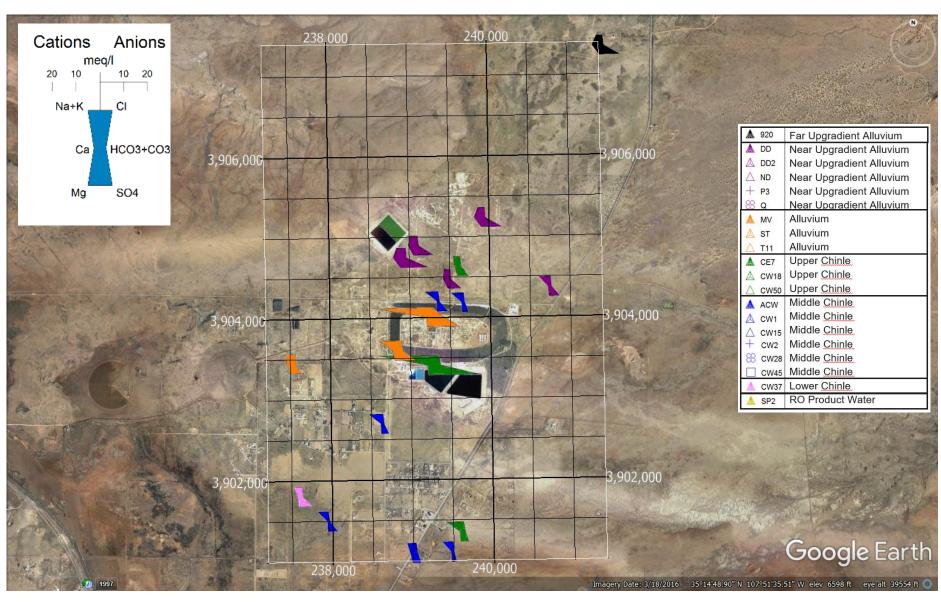
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Cations Anions meq/l 20 10 10 20 Na+K HCO3+CO3 Mg SO4

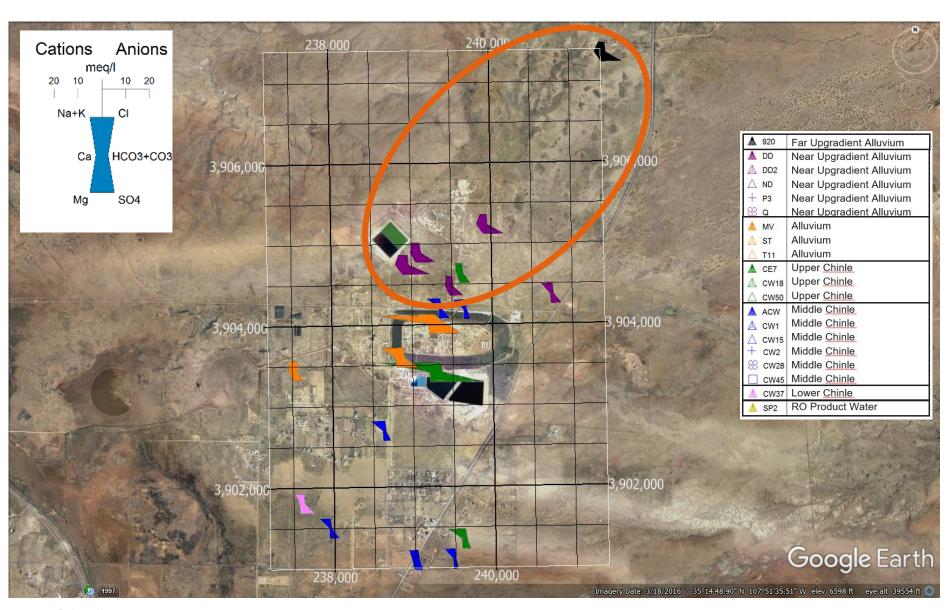




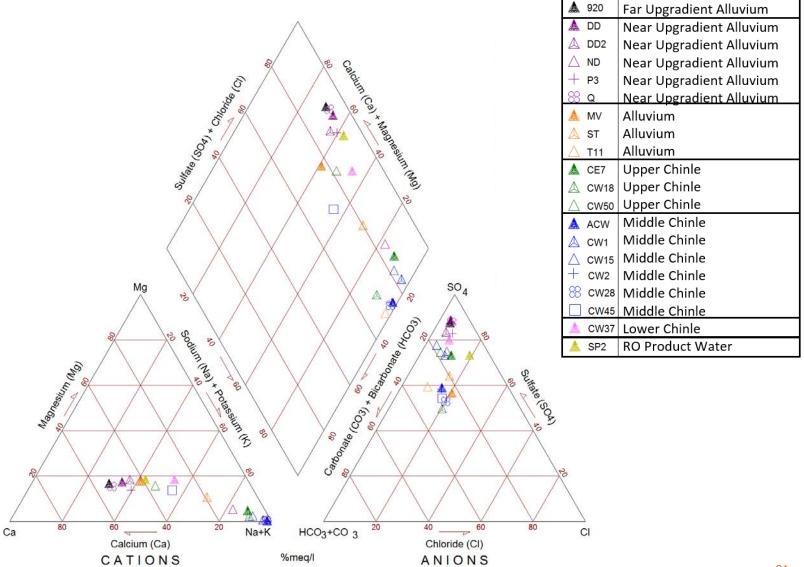








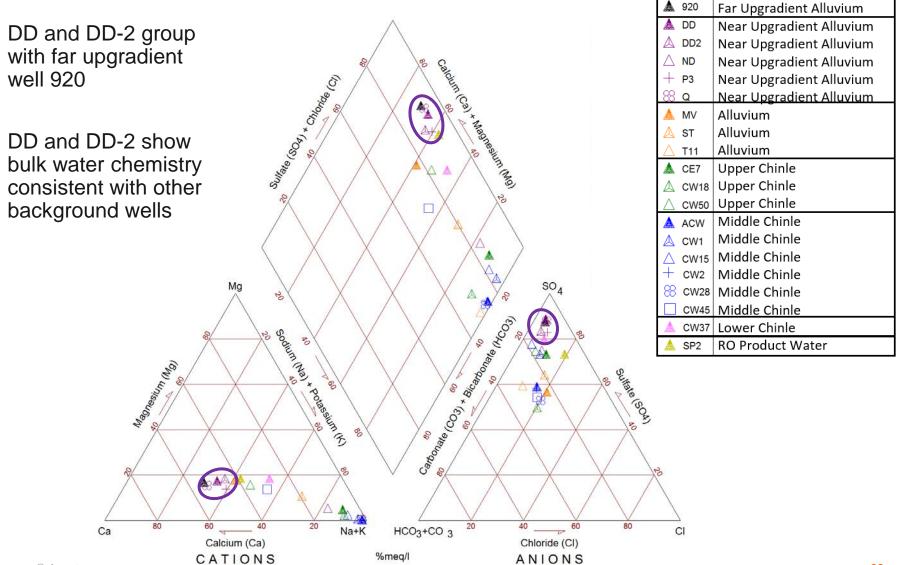




© Arcac

Piper Diagram: the standard for fingerprinting water







Split Sampling data – sampling methods



- Volumetric purge
 - 3 casing volumes
 - Parameter stability
- Micropurge: immediate collection of first water
- Passive samplers: collection of equilibrated water

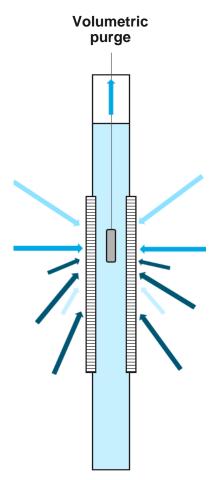




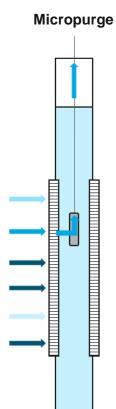


Color indicates concentration

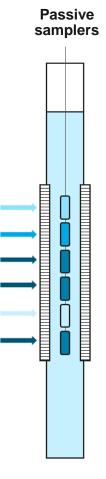
Length indicates transmissivity



- Direct sample of aquifer water
- 3D spatial average
- More transmissive zones dominate, but pulls from low transmissivity units
- Clears well of misrepresentative water prior to sampling



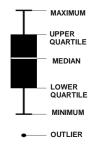
- Direct sample of well water at discrete depth
- If tight formation, sample is solely well water
- Should roughly equal passive sampler data at same depth



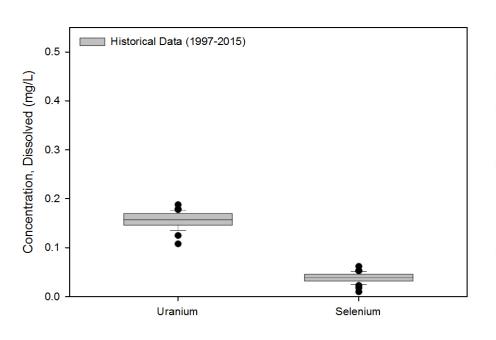
- Equilibrate with water in well
- Time-weighted average of all water through well over entire deployment (4 weeks)
- Theory: represents water flowing through formation at that discrete depth

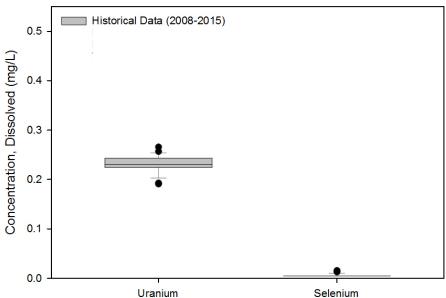
DD and DD-2 split sampling results compared to historical data





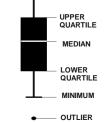
Well DD Well DD-2





DD and DD-2 split sampling results compared to historical data

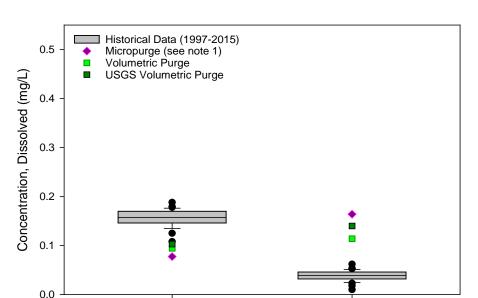




MAXIMUM

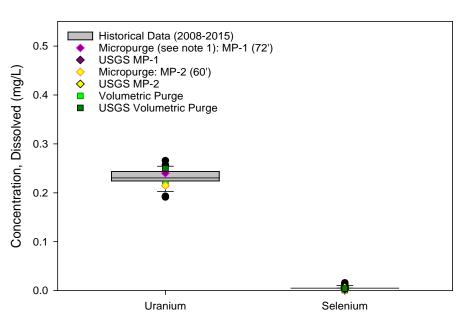
DD

Micropurge and volumetric purge showed lower uranium and higher selenium than is typical of well DD



DD-2

Micropurge and volumetric purge showed similar uranium and selenium as is typical of well DD-2



Notes.

1. Micropurge concentrations are total metals, not dissolved.

Uranium

Notes:

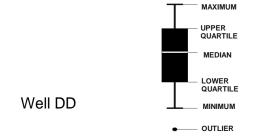
1. Micropurge concentrations are total metals, not dissolved.

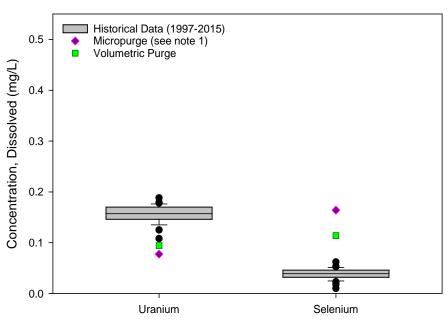
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Selenium

Results by sample method: PARCADIS Pesign & Consultancy for natural and built assets well DD







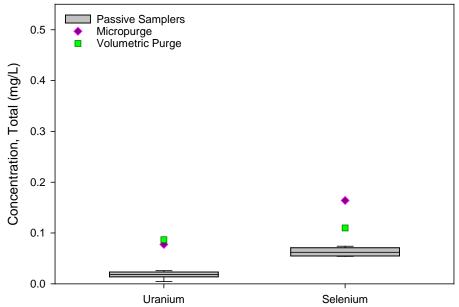
1. Micropurge concentrations are total metals, not dissolved.

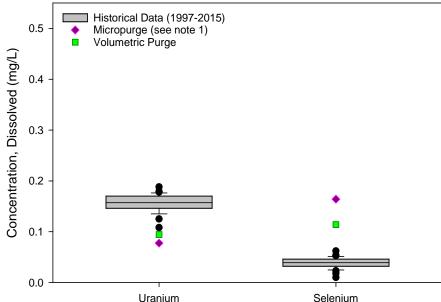
Results by sample method: PARCADIS Consultancy for natural and built assets well DD



MAXIMUM







Passive sampler results are much lower than either micropurge or volumetric purge

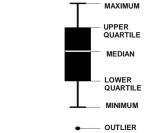
1. Micropurge concentrations are total metals, not dissolved.

Passive sampler results are also much lower than historical data

Results by sample method: PARCADIS Posign & Consultancy for natural and built assets well DD-2



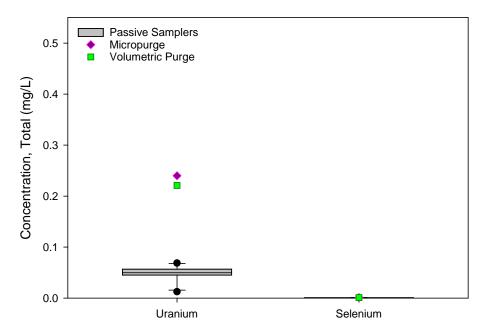
Well DD-2



Selenium

30

Well DD-2



0.0 Uranium

1. Micropurge concentrations are total metals, not dissolved.

Historical Data (2008-2015)

Micropurge (see note 1)

Volumetric Purae

0.5

0.3

0.2

0.1

Concentration, Dissolved (mg/L)

Passive sampler results are much lower than either micropurge or volumetric purge

Passive sampler results are also much lower than historical data



Closed square = volumetric purge

Open circle = micropurge

Closed circle, line = passive samplers



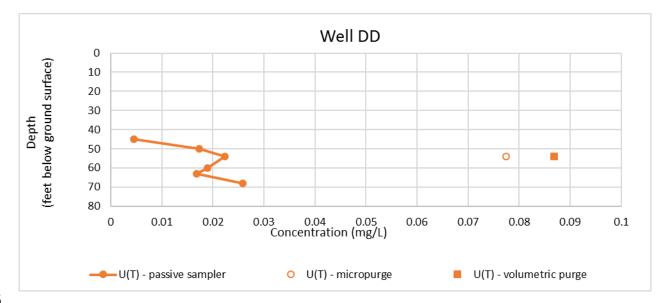
Closed square = volumetric purge

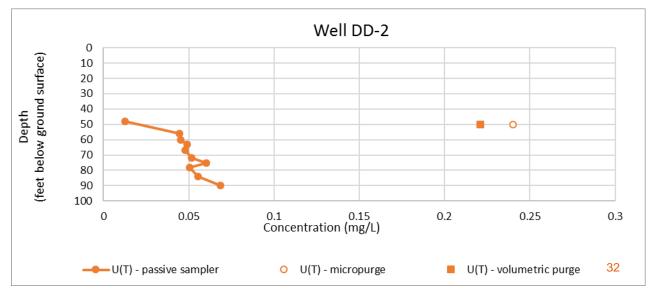
Open circle = micropurge

Closed circle, line = passive samplers

Passive sampler results are much lower than either micropurge or volumetric purge

Passive sampler results are not equivalent to micropurge results at same depth







Closed square = volumetric purge

Open circle = micropurge

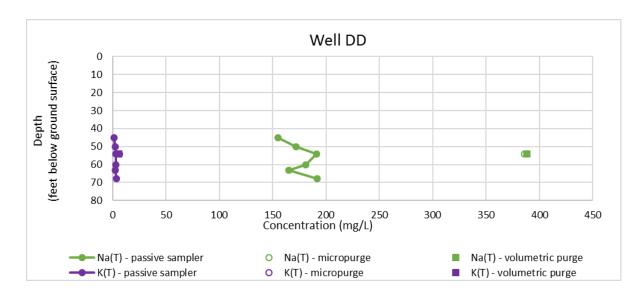
Closed circle, line = passive samplers

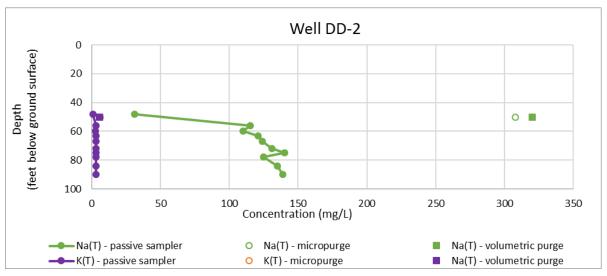
Green is sodium

Purple is potassium

Conservative ions did not equilibrate

Passive samplers did not collect a representative water sample





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Passive Samplers – bench testing



Passive sampler bench testing

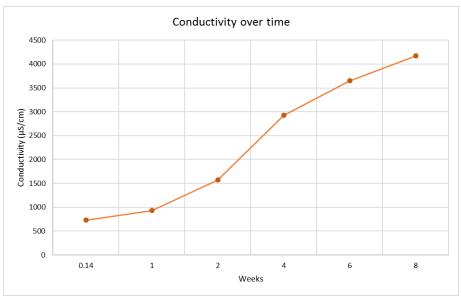
- Samplers used in this test were modeled after the samplers used in the split sampling
- Opted for setup that allowed as much free flow of water as possible
 - Did not use red netting or any other material that could restrict flow
- Tested key elements uranium and selenium
- Tested conservative ions sodium and potassium as controls
 - Should show maximum possible diffusion because these do not react

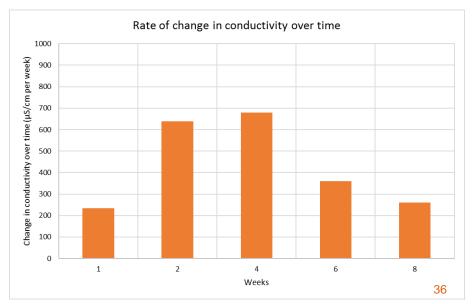
Passive sampler bench testing results

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 Conductivity was still increasing in the collected passive samplers at 8 weeks' time

- Peak change in conductivity in the passive samplers occurred at 4 weeks
- Reflects that highest mass flux was occurring around when passive samplers were collected in the field





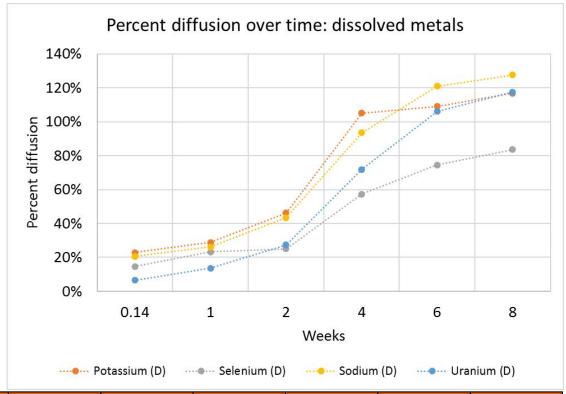
Passive sampler bench

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testing results

It takes <u>at least 6 weeks</u>, and likely 8+ weeks, for equilibration to occur

We also saw binding to the nylon mesh, including up to 5 mg/kg uranium



	Fully mixed solution	24hr	Week 1	Week 2	Week 4	Week 6	Week 8		
Analyte	% diffusion	% diffusion	% diffusion	% diffusion	% diffusion	% diffusion	% diffusion		
Dissolved Metals by Method SW6010B for Na and K, SW6020 for Se and U (mg/L)									
Potassium	100%	23%	29%	46%	105%	109%	117%		
Selenium	100%	15%	23%	25%	57%	75%	84%		
Sodium	100%	21%	26%	43%	94%	121%	128%		
Uranium	100%	6%	14%	27%	72%	106%	117%		



Geophysics

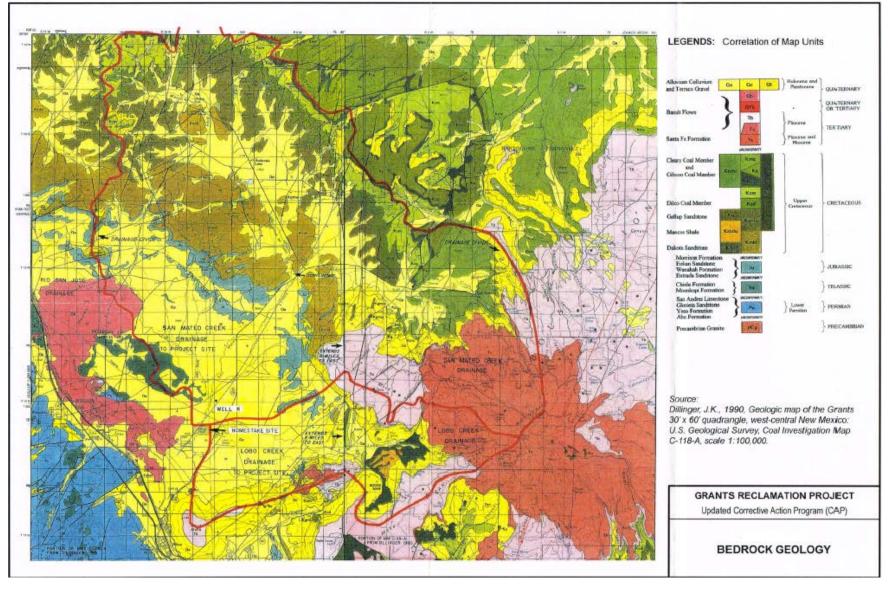
Logging Method Matrix ARCADIS Design & Consultancy for natural and built assets



Track	1	2	3	4	5
Content	Natural Gamma Ray	Uranium in parts per million (ppm)	Depth in feet	Lithology	Well Construction
	Induction Resistivity	Thorium/potassium (Th/K) Ratio in ppm/%			
Data Used to create Log	Natural Gamma Ray Induction Conductivity	Spectral Gamma Ray data processed into K, U and Th concentrations in picoCuries per gram (pCi/g)	Depth data gathered from each logging run	Natural Gamma, Induction Resistivity, Th/K Ratio, U concentration, Fluid Conductivity, Flowmeter Logs, Descriptive Logs	Optical Televiewer, Caliper, Fluid Temperature, Fluid Conductivity, Well construction documentation
Processing	Minimally processed data provided by USGS	K, U, and Th values recalculated ¹ to % or ppm	None	Experience based interpretation	Compilation of historic data and in- well observations from geophysical logging
Comments	Primary logs used to interpret lithology outside the borehole – See Track 4	Uranium plotted as ppm, reflective of presumably U content in matrix primarily. Could provide insight into uranium concentrations in groundwater. Th/K Ratio useful for ascertaining degree of weathering/maturity of sediments, supplemented by Track 4	Common depth reference (ground surface) used for all logging probes, essential for properly aligning various data tracks.	The composite interpretation of geophysical and descriptive log data were used to infer the lithological conditions. Used to create sections A-A' and B-B'	Data mainly provided the condition of the interior of the wells, and historic data used for annular space.

Bedrock Geology Map





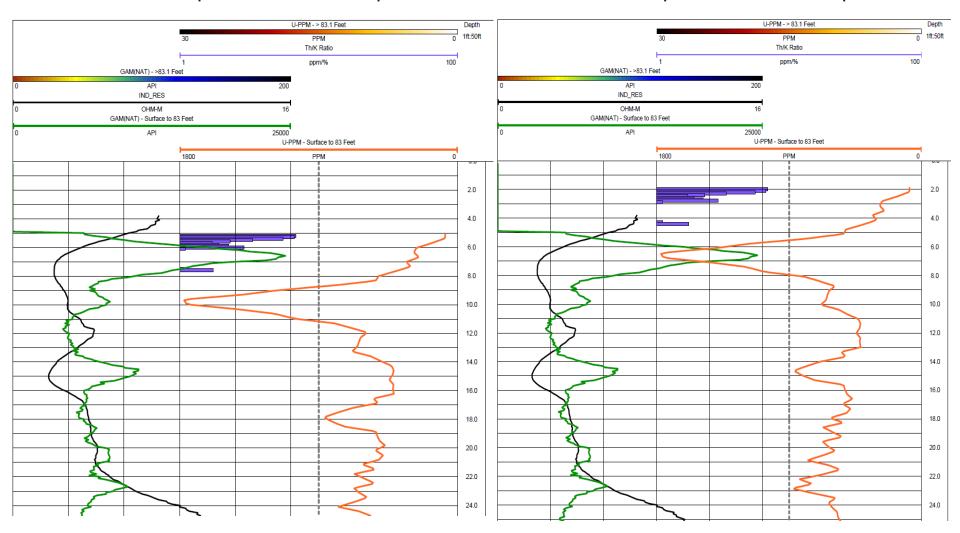




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Uncorrected Spectral Gamma Depth

Corrected Spectral Gamma Depth





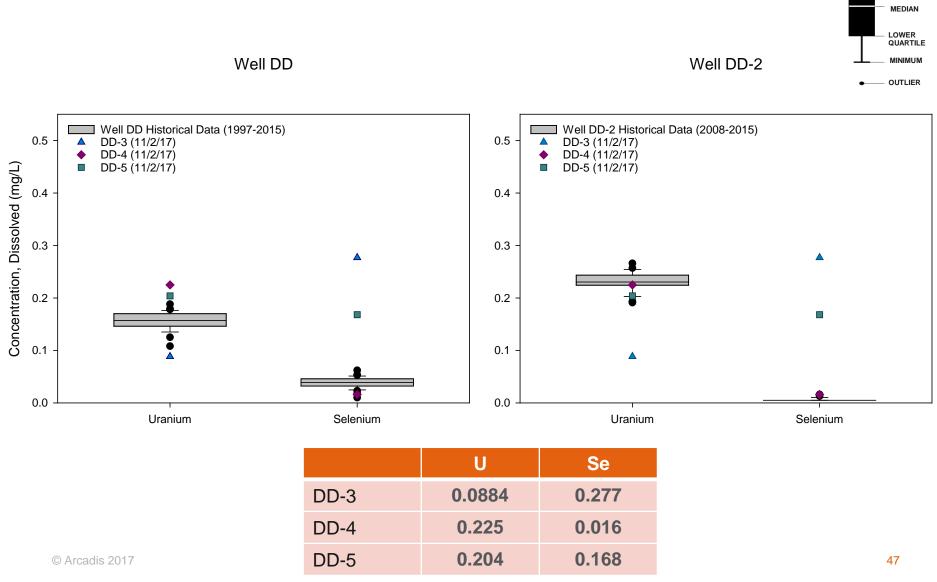
Wells DD-3, DD-4, and DD-5







MAXIMUM UPPER QUARTILE



DD-6 and DD-7 are dry



Drilling and geophysics at DD and DD-2

Location of boreholes





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Sampling and analysis

Samples collected covering both saturated and unsaturated zones

Sample location selection based on lithological characteristics and on dynamic spectral gamma data

Static spectral gamma collected at each sampling location

Analyses

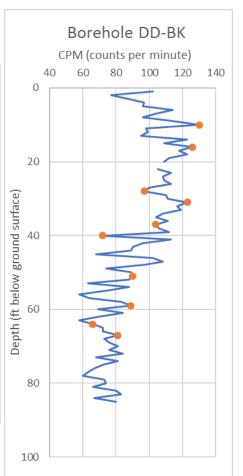
- Total metals
- Alkaline leaching test (modified SPLP based on Kohler et al. 2004)
- Particle size analysis
- Microscopic and spectroscopic analysis

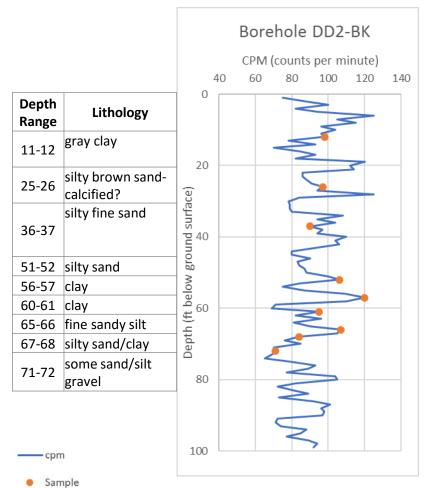




Geologic logs

Depth Range	Lithology
9-10	clay w/ trace sand
15-16	fine to medium sand with some silt
27-28	fine to coarse sand, trace silt
30-31	silty sand and gravel
36-37	clay
39-40	silt fine sand with hard layer
50-51	silty fine to coarse sand
58-59	clay
63-64	silty fine sand
66-67	silty fine sand



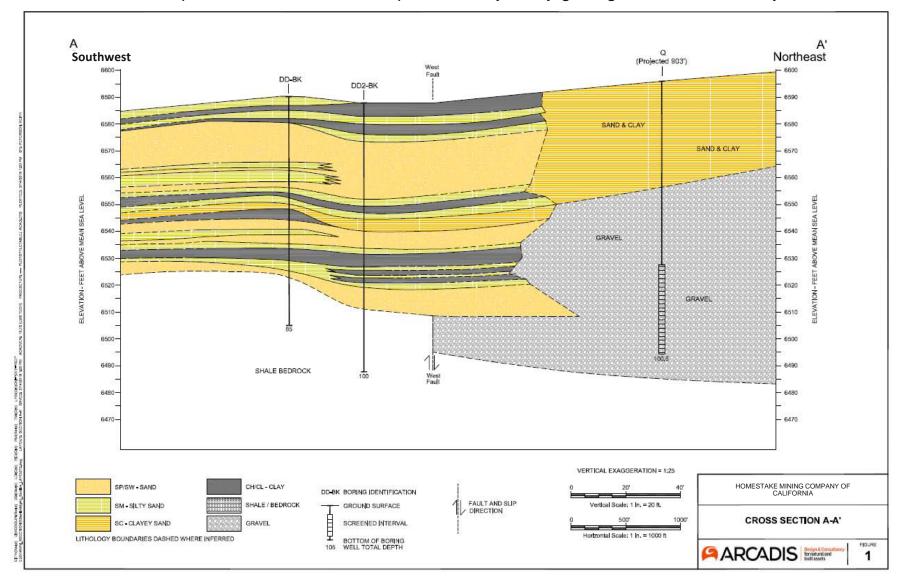


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Cross section



New information in this area! Changes the overall interpretation of the DD/DD-2 area and is more consistent with depositional environment as presented by many geologists over last 100+ years





Lab results and report

- Lab results are expected this week for the total metals, leachable metals, and particle size analysis
- Samples will then be selected for microscopy and spectroscopy based on geochemical results
- If data are received in a timely manner, report could be forth-coming by mid-May